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Write the nuclear equation for this reaction and identify the two other particles. ${}_{11}^{23}\text{Na} + {}_0^1\text{n} \rightarrow {}_{11}^{24}\text{Na} + {}_0^1\text{n}$
The two other particles formed are neutrons. 8. When bombarded with neutrons, lithium-6 produced an alpha particle and an isotope of hydrogen.

KEY - Nuclear Equations

Defining key concepts - ensure that you can accurately define main terms, ... Problem solving - apply what you've learned to solve nuclear equation practice problems Additional Learning.

Quiz & Worksheet - How to Balance Nuclear Equations ...

Write and balance nuclear equations Changes of nuclei that result in changes in their atomic numbers, mass numbers, or energy states are nuclear reactions . To describe a nuclear

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reaction, we use an equation that identifies the nuclides involved in the reaction, their mass numbers and atomic numbers, and the other particles involved in the ...

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A balanced nuclear reaction equation indicates that there is a rearrangement during a nuclear reaction, but of subatomic particles rather than atoms. Nuclear reactions also follow conservation laws, and they are balanced in two ways: The sum of the mass numbers of the reactants equals the sum of the mass numbers of the products.

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