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...

Matter is conserved in
any physical or
chemical change- the
total mass of the
substance before the

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Answer Key
change (product) is always equal to the total mass of the substance after the change (reactant).

**Chemistry Chapters
1 & 2 Test: Matter
and Change ...**

Matter and Change.

SECTION 3. SHORT

ANSWERS

Answer the following questions in the space provided. 1.

A horizontal row of elements in the periodic table is called

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a(n) period . 2. The symbol for the element in Period 2, Group 13, is B . 3. Elements that are good conductors of heat and electricity are metals .

1 Matter and Change - HUBBARD'S

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ANSWERS Chemistry:
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Answers 8 9. The first
ionization energies

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generally increase as you move left-to-right across a period. The increased nuclear charge of each successive element produces an increased hold on the valence electrons. 10. The first ionization energies generally decrease as

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answers

Supplemental Problems

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Matter—Properties and

Changes

Changes 1. An 18-g sample of element A combines completely with a 4-g sample of element B to form the compound AB. What is the mass of the compound formed? 2. A substance breaks down into three component elements when it is heated.

Supplemental

Page 10/27

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Problems - MARRIC

chemistas someone who studies matter and the changes that it undergoes. The word dependable is described by the synonyms reliable and trusted. The setting of the sentence and the action describe a situation that is positive and full of celebration. The elements that are mentioned are all gases.

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Solutions Manual

Chemistry: Matter and
Change • Chapter 5 73

CHAPTER 5 SOLUTIONS
MANUAL Chapter 5

Assessment pages

166–169 Section 5.1

Mastering Concepts 34.

Define the following
terms. a. frequency

Frequency is the
number of waves that
pass a given point per

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second. b. wavelength
Wavelength is the
shortest distance
between

Chapter 5
Assessment

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Page v. Research-
Based Vocabulary
Development Third,
you will notice that
vocabulary is
introduced and

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practiced throughout
the Science Notebook.

When students know
the meaning of the
words used to discuss
information, they are

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Answers will vary but
should demonstrate an
understanding of the
need to experiment
and examine data and
conditions carefully.

(10/4) ... chemistry to

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artificially duplicate
normal body. functions.

6. heart pacemaker
and defibrillator ... This
joining of atoms is
called fusion, and it
changes matter to the
energy that powers the
star. 7. We ...

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Terms in this set (38)

chemistry. the study of

the composition,

structure, and

properties of matter

and the changes it

undergoes. 6 Branches

of Chemistry. Organic

Inorganic Physical

Analytical

Study 38 Terms |

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apparent changes in matter result from changes in the groupings of atoms. Democritus further believed that matter is composed of empty space through which atoms move, that different kinds of atoms come in different sizes and shapes, and that the differing properties of atoms are due to the size, shape, and movement of the

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**The Structure of the
AtomThe Structure
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Characteristics Use

with pages 9—10.

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II. Chemistry: Matter

and Change (Details—n

Compare and contrast

mass and weight using

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the Venn diagram below does not reflect gravitational pull on matter a measure of the effect of gravitational pull on matter

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graded to find out
where you took a
wrong turn.

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Answers Key

**Chemistry 1
Worksheet
Classification Of
Matter And Changes**

...

A chemical change is any change that results in the formation of new chemical substances. At the molecular level, chemical change involves making or breaking of bonds between atoms. These changes are chemical:

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Ex. iron rusting (iron oxide forms) A physical change is any change NOT involving a change in the substance's chemical identity. Here are some examples:

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chemistry: matter and

change ? A 24-g

sample of carbon

combines with 64 g of

oxygen to form CO₂.

what is the mass of the

reactants? what is the

mass of the product?

which law do these

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extraordinary points.

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