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Chapter 5 - Electrons in Atoms - 5 Assessment - Page 154: 80 Answer Electrons occupy equal energy orbitals singly before beginning to pair up, because according to Hund's rule the most stable arrangement of electrons occurs when electrons with the same spin is as large as possible.

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It is important, therefore, to understand the arrangement of an atom's electrons. This 10-question multiple-choice chemistry practice test deals with the concepts of electronic structure, Hund's Rule, quantum numbers, and the Bohr atom. Answers to the questions appear at the end of the test.

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