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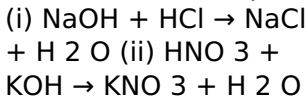
Answer-1. Post-Your-
Explanation-1. 2. Bases
react with. acids to
produce salts and
water. water to
produce acids and
salts. salts to produce
acids and water.
neither acids, salts, nor
water. Answer-2.

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Acids Bases and Salts Multiple Choice Questions and Answers

A reaction in which an acid and base react with each other to give a salt and water is termed as neutralization

reaction. For Example:



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Short Answer type

Question 10 Give a reason why (a) Great care should be taken while handling laboratory acids and bases. (b) Solid powder of baking soda does not turn dry litmus paper to blue in color.

Question 11 Define acid rain and how it is produced? What are its harmful effect?

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Question 12 Define neutralization reaction with an example.

Worksheet for Class 7 Science Chapter 5: Acid Base and Salt

Answer. (a) HCl is acid and NaOH is base whose combination forms the common salt. Its formula is NaCl (Sodium chloride). It is obtained from sea water. (b) Rock salt is the common name for the mineral "halite". Its

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chemical formula is NaCl. It may be white or light blue or yellow depending upon impurities present in it.

Acids Bases and Salts Chapter Wise Important Questions

...

Here both acids and bases get neutralised For example, when sodium hydroxide (NaOH) is added to hydrochloric acid (HCl), sodium chloride (NaCl)

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and water (H_2O) are obtained. $NaOH + HCl \rightarrow NaCl + H_2O + \text{Heat}$
6. Mark 'T' if the statement is true and 'F' if it is false:

NCERT Solutions Class 7 Science Chapter 5 Acid Base and ...

An acid is a substance that reacts with water to produce hydronium ions, H_3O^+ . A base is a substance that produces hydroxide

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ions, OH^- , in water.

CHEM 0012 Lecture
Notes 3. When an acid
and base reacts, an
ionic compound is
produced as one of the
products. We call ionic
compounds "salts".

Acids, Bases and Salts

The salts of strong
acids and weak bases
give acidic solution
having pH less than 7.
Example, NH_4Cl ,
Ammonium Chloride

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will have pH less than 7. The salts of weak acids and strong bases give basic solution having pH more than 7. Example, Na_2CO_3 , Sodium Carbonate will have pH more than 7.
Question 7.

Acids, Bases and Salts Class 10 Important Questions with ...

neutralization reaction.
a reaction in which an acid and a base react

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in an aqueous solution to produce a salt and water (19.4) pH. a number used to denote the hydrogen-ion concentration, or acidity, of a solution; it is the negative logarithm of the hydrogen-ion concentration of a solution (19.2) salt hydrolysis.

Chapter 19 Acids, Bases, and Salts Flashcards | Quizlet

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Which of the following
values would represent
the pH of a strong
base?

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Important Questions for CBSE Class 10 Science Chapter 2 ...

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question_answer22) A
base reacts with an
acid to form salt and

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water. What is this reaction called? A) Oxidation reaction done clear. B) Neutralization reaction done clear. C) Reduction reaction done clear. D) Biochemical reaction done clear. View Solution  question_answer23) Identify the base used in making antacids.

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